## organic compounds

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### 5,11,17,23-Tetra-tert-butyl-25,27-bis[2-(4-nitrophenoxy)ethoxy]calix[4]arene-26,28-diol acetonitrile tetrasolvate

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Key indicators: single-crystal X-ray study; T = 173 K; mean  $\sigma$ (C–C) = 0.008 Å; disorder in main residue; R factor = 0.114; wR factor = 0.349; data-to-parameter ratio = 14.4.

In the crystal structure of the title compound,  $C_{60}H_{70}N_2O_{10}$ . 4CH<sub>3</sub>CN, the calix[4]arene molecule adopts an open-cone conformation with two intramolecular O-H···O hydrogen bonds. The four benzene rings of the calix[4]arene are twisted to the mean plane defined by four methylene C atoms bridging the benzene rings, with dihedral angles ranging from 57.74 (10) to 65.99 (12)°. Two pendant nitrophenyl rings are nearly perpendicular to each other, the dihedral angle being 70.9 (3) $^{\circ}$ . The asymmetric unit of the crystal structure contains four acetonitrile solvent molecules, one of which lies in the calix cavity and makes  $C-H \cdots \pi$  interactions and another links with the calix [4] arene via  $C-H \cdots O$  hydrogen bonding. One tert-butyl group is disordered over two sets of sites, with a 0.736 (13):0.264 (13) occupancy ratio.

### **Related literature**

For general background to the chemistry of calix[4]arenes, see: Gutsche (1998). For related crystal structures, see: Singh et al. (2004); Bolte et al. (2003); Zeng et al. (2002); Gale et al. (1998); Drew et al. (1997); Böhmer et al. (1993); Bugge et al. (1992). For C-H··· $\pi$  contacts, see: Tsuzuki *et al.* (2000); Umezawa *et* al. (1998). For inclusion complexes, see: McKervey et al. (1986).



### **Experimental**

### Crystal data C60H70N2O10.4C2H2N

$\gamma = 82.230 \ (3)^{\circ}$
$V = 3270.5 (13) \text{ Å}^3$
Z = 2
Mo $K\alpha$ radiation
$\mu = 0.08 \text{ mm}^{-1}$
T = 173  K
$0.51 \times 0.35 \times 0.15 \text{ mm}$

### Data collection

Bruker SMART CCD area-detector diffractometer Absorption correction: none 15816 measured reflections

#### Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.114$  $wR(F^2) = 0.349$ S = 1.0511155 reflections 775 parameters

### Table 1

Hydrogen-bond geometry (Å, °).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdots A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
O1-H1···O3	0.82	1.99	2.802 (6)	171
O2−H2···O4	0.82	1.97	2.781 (4)	176
C50−H50C···O7 <sup>i</sup>	0.96	2.59	3.487 (10)	156
C68−H68B····O10	0.96	2.39	3.247 (13)	148
$C65-H65A\cdots Cg1$	0.96	2.66	3.590 (6)	163

Symmetry code: (i) -x + 1, -y + 1, -z + 1. Cg1 is the centroid of the C42–C47 ring.

Data collection: SMART (Bruker, 1999); cell refinement: SAINT (Bruker, 1999); data reduction: SAINT; program(s) used to solve structure: SHELXTL (Sheldrick, 2008); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: XU2512).

11155 independent reflections

7114 reflections with  $I > 2\sigma(I)$ 

H-atom parameters constrained

 $R_{\rm int} = 0.049$ 

5 restraints

 $\Delta \rho_{\rm max} = 0.78 \ {\rm e} \ {\rm \AA}^-$ 

 $\Delta \rho_{\rm min} = -0.64 \text{ e} \text{ Å}^{-3}$ 

### References

- Böhmer, V., Ferguson, G., Gallagher, J. F., Lough, A. J., McKervey, M. A., Madigan, E., Moran, M. B., Phillips, J. & Williams, J. (1993). J. Chem. Soc. Perkin Trans. 1, pp. 1521–1527.
- Bolte, M., Danila, C. & Böhmer, V. (2003). Acta Cryst. E59, 0533-0534.
- Bruker (1999). SMART and SAINT. Bruker AXS Inc., Madison, Wisconsin, USA.
- Bugge, K.-E., Verboom, W., Reinhoudt, D. N. & Harkema, S. (1992). Acta Cryst. C48, 1848–1851.
- Drew, M. G. B., Beer, P. D. & Ogden, M. I. (1997). Acta Cryst. C53, 472–474.
  Gale, P. A., Chen, Z., Drew, M. G. B., Heath, J. A. & Beer, P. D. (1998). Polyhedron, 17, 405–412.
- Gutsche, C. D. (1998). *Calixarenes Revisited*. RSC Monographs in Chemistry. Cambridge: Royal Society of Chemistry.
- McKervey, M. A., Seward, E. M., Ferguson, G. & Ruhl, B. L. (1986). J. Org. Chem. 51, 3581–3584.
- Sheldrick, G. M. (2008). Acta Cryst. A64, 112-122.
- Singh, N., Kumar, M. & Hundal, G. (2004). Tetrahedron 60, 5393-5405.
- Tsuzuki, S., Honda, K., Uchimaru, T., Mikami, M. & Tanabe, K. (2000). J. Am. Chem. Soc. 122, 3746–3753.
- Umezawa, Y., Tsuboyama, S., Honda, K., Uzawa, J. & Nishio, M. (1998). Bull. Chem. Soc. Jpn, **71**, 1207–1213.
- Zeng, X.-S., Weng, L.-H., Chen, L.-X., Xu, F.-B., Li, Q.-S., Leng, X.-B., He, X.-W. & Zhang, Z.-Z. (2002). *Tetrahedron*, **58**, 2647–2658.

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# 5,11,17,23-Tetra-*tert*-butyl-25,27-bis[2-(4-nitrophenoxy)ethoxy]calix[4]arene-26,28-diol acetonitrile tetrasolvate

### J.-M. Yuan, Y.-H. Gao, J.-P. Ma and D.-S. Guo

### Comment

Calix[4]arenes have been attracting much interest because they possess a versatile three-dimensional cavity and are ideal scaffolds for the construction of supramolecular systems. In particular, the lower or upper rim of a calix[4]arene platform can be modified to achieve more sophisticated receptors with a specific affinity and selectivity for ion recognition (Gutsche, 1998). Several crystal structures of 1,3-substituted cone calix[4]arene derivatives (Singh *et al.*, 2004; Bolte *et al.*, 2003; Zeng *et al.*, 2002; Gale *et al.*, 1998; Drew *et al.*, 1997; Böhmer *et al.*, 1993; Bugge *et al.*, 1992) have been described. We report here the crystal structure of a new 1,3-substituted calix[4]arene,  $C_{60}H_{70}N_2O_{10}.4CH_3CN$ , namely 5,11,17,23-tetra-*tert*-butyl-25,27-bis[2-(4-nitrophenoxy)ethoxy]-26,28-dihydroxycalix[4]arene acetonitrile tetrasolvate.

In the crystal structure of the title compound, as shown in Fig. 1, the molecule of the calix[4]arene adopts an opencone conformation, in which either phenol hydroxy group links with one neighboring ethereal O atom *via* an intramolecular O—H···O hydrogen bond (Table 1) and one *t*-butyl group shows rotational disorder. The four benzene rings of the calix[4]arene are twisted to the virtual plane defined by four methylene C atoms bridging the phenolic rings with dihedral angles ranging from 57.74 (10) to 65.99 (12)°. Two pendant nitrophenyl rings are nearly perpendicular to each other, the dihedral angle being 70.9 (3)°. This conformation results in a distance of 4.005 (6) Å between diametrically opposed atoms O1 and O2, almost same as 4.289 (6) Å between O3 and O4.

There are four acetonitrile solvate molecules in the asymmetric unit of the crystal structure, one of which lies in the calix cavity with C—H<sup>...</sup> $\pi$  contacts (Umezawa *et al.*, 1998; Tsuzuki *et al.*, 2000) and another links with the calix[4]arene *via* C—H<sup>...</sup>O hydrogen bonding (Table 1). The apolar end of the acetonitrile is held in the cavity, while the polar one remains outside, similar to the related cone calix[4]arene system (McKervey *et al.*, 1986) where an acetonitrile molecule is included. The intermolecular C—H<sup>...</sup>O hydrogen bonds (Table 1) and the remaining acetonitrile molecules stabilize the molecular packing.

### Experimental

To a refluxing suspension of *p-tert*-butylcalix[4]arene (1.112 g, 1.50 mmol) and anhydrous potassium carbonate (0.228 g, 1.65 mmol) in dry acetonitrile (15 ml) was added 2-(4-nitrophenoxy)ethyl-4-methylbenzenesulfonate (1.144 g, 3.00 mmol) in dry acetonitrile (15 ml) dropwise. The mixture was stirred and refluxed under a nitrogen atmosphere for 46 h and cooled to room temperature. The solvent was removed under reduced pressure. The residue was neutralized with diluted hydrochloric acid and extracted with dichloromethane. The organic layer was washed with saturated sodium hydrogen carbonate and brine, and dried over anhydrous magnesium sulfate. Removal of the solvent under reduced pressure, the residue was purified by flash column chromatography (silica gel, ethyl acetate/hexane/dichloromethane = 1:12:4,  $R_F = 1/2$ ) to give the title compound in 88% yield as a white solid, m.p. 397–399 K. Single crystals suitable for X-ray diffraction analysis were obtained from slow evaporation of a solution in acetonitrile at 298 K.

### Refinement

All H atoms were placed in geometrically idealized positions and refined using a riding model, with C—H distances of 0.93–0.97 Å, and with  $U_{iso}(H)$  values of  $1.5U_{eq}(C)$  for methyl H atoms and  $1.2U_{eq}(C)$  for the other H atoms. In the title compound, one *tert*-butyl group (C1–C4) is rotational disordered over two sites; the site-occupancies were refined to 0.736 (13):0.264 (13). The C—C bond lengths involving the disordered atoms were restrained to be similar. The C67—C68 and C67—N6 bond lengths were restrained. As the quality of the crystal is poor the accuracy of the determination is low.

### **Figures**



Fig. 1. The molecular structure of the title compound, with displacement ellipsoids drawn at the 30% probability level. The minor disordered component and H atoms have been omitted for clarity.

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### Crystal data

$C_{60}H_{70}N_2O_{10}\cdot 4C_2H_3N$	<i>Z</i> = 2
$M_r = 1143.40$	$F_{000} = 1224$
Triclinic, $PT$	$D_{\rm x} = 1.161 {\rm Mg m}^{-3}$
Hall symbol: -P 1	Mo $K\alpha$ radiation $\lambda = 0.71073$ Å
<i>a</i> = 13.195 (3) Å	Cell parameters from 4717 reflections
b = 13.388 (3)  Å	$\theta = 2.6 - 27.2^{\circ}$
c = 19.692 (5)  Å	$\mu = 0.08 \text{ mm}^{-1}$
$\alpha = 71.978 \ (3)^{\circ}$	T = 173  K
$\beta = 84.022 \ (3)^{\circ}$	Block, colourless
$\gamma = 82.230 \ (3)^{\circ}$	$0.51 \times 0.35 \times 0.15 \text{ mm}$
$V = 3270.5 (13) \text{ Å}^3$	

### Data collection

Bruker SMART CCD area-detector diffractometer	7114 reflections with $I > 2\sigma(I)$
Radiation source: fine-focus sealed tube	$R_{\rm int} = 0.049$
Monochromator: graphite	$\theta_{\text{max}} = 25.0^{\circ}$
<i>T</i> = 173 K	$\theta_{\min} = 1.6^{\circ}$

$\phi$ and $\omega$ scans	$h = -15 \rightarrow 15$
Absorption correction: none	$k = -15 \rightarrow 15$
15816 measured reflections	$l = -13 \rightarrow 23$
11155 independent reflections	

### Refinement

Refinement on $F^2$	Secondary atom site location: difference Fourier map
Least-squares matrix: full	Hydrogen site location: inferred from neighbouring sites
$R[F^2 > 2\sigma(F^2)] = 0.114$	H-atom parameters constrained
$wR(F^2) = 0.349$	$w = 1/[\sigma^2(F_o^2) + (0.1473P)^2 + 8.6696P]$ where $P = (F_o^2 + 2F_c^2)/3$
<i>S</i> = 1.05	$(\Delta/\sigma)_{\text{max}} = 0.005$
11155 reflections	$\Delta \rho_{max} = 0.78 \text{ e} \text{ Å}^{-3}$
775 parameters	$\Delta \rho_{min} = -0.64 \text{ e } \text{\AA}^{-3}$
5 restraints	Extinction correction: none
Primary atom site location: structure-invariant direct	

Primary atom site location: structure-invariant direct methods

### Special details

**Experimental**. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>): δ 8.19 (d, 4H, *J* = 9.20 Hz), 7.09 (s, 2H), 7.06 (s, 4H), 6.99 (d, 4H, *J* = 9.20 Hz), 6.82 (s, 4H), 4.35 (s, 8H), 4.33 (d, 4H, *J* = 13.02 Hz), 3.31 (d, 4H, *J* = 13.02 Hz), 1.29 (s, 18H), 0.98 (s, 18H).

**Geometry**. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement**. Refinement of  $F^2$  against ALL reflections. The weighted *R*-factor *wR* and goodness of fit *S* are based on  $F^2$ , conventional *R*-factors *R* are based on *F*, with *F* set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating *R*-factors(gt) *etc*. and is not relevant to the choice of reflections for refinement. *R*-factors based on  $F^2$  are statistically about twice as large as those based on *F*, and *R*- factors based on ALL data will be even larger.

	x	у	Ζ	$U_{\rm iso}*/U_{\rm eq}$	Occ. (<1)
C1	0.2547 (7)	0.3466 (10)	0.8947 (5)	0.069 (3)	0.739 (13)
H1A	0.2256	0.3367	0.9428	0.104*	0.739 (13)
H1B	0.2453	0.2872	0.8795	0.104*	0.739 (13)
H1C	0.2213	0.4099	0.8635	0.104*	0.739 (13)
C2	0.3929 (11)	0.4421 (10)	0.9200 (6)	0.072 (4)	0.739 (13)
H2A	0.4656	0.4430	0.9201	0.107*	0.739 (13)
H2B	0.3626	0.4301	0.9679	0.107*	0.739 (13)
H2C	0.3636	0.5088	0.8901	0.107*	0.739 (13)
C3	0.4147 (10)	0.2469 (7)	0.9405 (5)	0.072 (4)	0.739 (13)
H3A	0.4877	0.2435	0.9416	0.107*	0.739 (13)

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters  $(A^2)$ 

H3B	0.3988	0.1930	0.9217	0.107*	0.739 (13)
H3C	0.3834	0.2361	0.9881	0.107*	0.739 (13)
C4	0.3731 (5)	0.3569 (5)	0.8920 (3)	0.0485 (15)	
C5	0.4236 (4)	0.3688 (4)	0.8157 (3)	0.0390 (13)	
C6	0.3685 (4)	0.3745 (4)	0.7580 (3)	0.0370 (12)	
H6	0.2978	0.3740	0.7649	0.044*	
C7	0.4149 (4)	0.3810 (4)	0.6897 (3)	0.0306 (11)	
C8	0.5219 (4)	0.3828 (4)	0.6790 (2)	0.0303 (11)	
C9	0.5803 (4)	0.3783 (4)	0.7355 (3)	0.0322 (11)	
C10	0.5309 (4)	0.3711 (4)	0.8028 (3)	0.0363 (12)	
H10	0.5698	0.3677	0.8404	0.044*	
C11	0.6975 (4)	0.3713 (4)	0.7274 (3)	0.0342 (11)	
H11A	0.7194	0.4094	0.6787	0.041*	
H11B	0.7218	0.4040	0.7594	0.041*	
C12	0.7438 (3)	0.2563 (4)	0.7448 (3)	0.0293 (11)	
C13	0.7655 (3)	0.2053 (4)	0.6906 (2)	0.0275 (10)	
C14	0.7975 (3)	0.0963 (4)	0.7087 (3)	0.0285 (10)	
C15	0.8073 (3)	0.0420 (4)	0.7807 (3)	0.0315 (11)	
H15	0.8274	-0.0303	0.7932	0.038*	
C16	0.7887 (4)	0.0896 (4)	0.8355 (3)	0.0335 (11)	
C17	0.7542 (4)	0.1987 (4)	0.8148 (3)	0.0326 (11)	
H17	0.7381	0.2324	0.8500	0.039*	
C18	0.8001 (4)	0.0302 (4)	0.9147 (3)	0.0416 (13)	
C19	0.6944 (5)	0.0345 (6)	0.9561 (3)	0.0621 (18)	
H19A	0.6659	0.1069	0.9476	0.093*	
H19B	0.7022	0.0024	1.0064	0.093*	
H19C	0.6494	-0.0029	0.9400	0.093*	
C20	0.8751 (5)	0.0844 (6)	0.9452 (3)	0.0568 (17)	
H20A	0.9420	0.0787	0.9214	0.085*	
H20B	0.8790	0.0502	0.9956	0.085*	
H20C	0.8502	0.1577	0.9371	0.085*	
C21	0.8431 (6)	-0.0850(5)	0.9263 (3)	0.072 (2)	
H21A	0.7986	-0.1193	0.9073	0.107*	
H21B	0.8476	-0.1200	0.9766	0.107*	
H21C	0.9101	-0.0883	0.9024	0.107*	
C22	0.8396 (4)	0.2765 (4)	0.5701 (3)	0.0354 (12)	
H22A	0.8345	0.2354	0.5380	0.042*	
H22B	0.9016	0.2491	0.5953	0.042*	
C23	0.8444 (4)	0.3907 (4)	0.5278 (3)	0.0322 (11)	
H23A	0 8990	0 3969	0 4902	0.039*	
H23B	0 7802	0.4200	0 5058	0.039*	
C24	0.8708 (3)	0.5528 (4)	0.5498 (3)	0.0313(11)	
C25	0.8700(5)	0.5526(1) 0.6058(5)	0.6011 (3)	0.0315(11) 0.0485(15)	
H25	0 8677	0 5678	0 6494	0.058*	
C26	0.8784 (5)	0 7130 (5)	0 5813 (3)	0.0556 (16)	
H26	0.8787	0.7473	0.6157	0.067*	
C27	0.8846 (4)	0 7684 (4)	0.5092 (3)	0.0388 (12)	
C28	0 8843 (4)	0 7170 (4)	0.3092(3)	0.0366(12)	
H28	0.8000	0.7550	0.4000	0.044*	
1120	0.0700	0.7550	0.4022	0.044	

C29	0.8755 (4)	0.6102 (4)	0.4779 (3)	0.0334 (11)
H29	0.8727	0.5769	0.4432	0.040*
C30	0.8120 (3)	0.0326 (4)	0.6552 (3)	0.0295 (10)
H30A	0.8727	-0.0178	0.6652	0.035*
H30B	0.8221	0.0799	0.6071	0.035*
C31	0.7182 (3)	-0.0269 (4)	0.6594 (2)	0.0280 (10)
C32	0.6346 (4)	0.0221 (4)	0.6168 (2)	0.0289 (10)
C33	0.5466 (4)	-0.0287 (4)	0.6241 (2)	0.0307 (11)
C34	0.5401 (4)	-0.1265 (4)	0.6752 (3)	0.0368 (12)
H34	0.4806	-0.1592	0.6805	0.044*
C35	0.6199 (4)	-0.1770 (4)	0.7186 (3)	0.0370 (12)
C36	0.7094 (4)	-0.1248 (4)	0.7083 (3)	0.0324 (11)
H36	0.7644	-0.1581	0.7358	0.039*
C37	0.6147 (5)	-0.2837 (4)	0.7760 (3)	0.0445 (14)
C38	0.6961 (6)	-0.3656 (5)	0.7568 (4)	0.0629 (19)
H38A	0.6818	-0.3745	0.7123	0.094*
H38B	0.6947	-0.4319	0.7938	0.094*
H38C	0.7626	-0.3417	0.7522	0.094*
C39	0.6358 (6)	-0.2736 (5)	0.8489 (3)	0.0618 (18)
H39A	0.7029	-0.2519	0.8461	0.093*
H39B	0.6321	-0.3406	0.8849	0.093*
H39C	0.5855	-0.2219	0.8611	0.093*
C40	0.5086 (5)	-0.3241 (6)	0.7832 (4)	0.068 (2)
H40A	0.4562	-0.2715	0.7918	0.103*
H40B	0.5065	-0.3879	0.8225	0.103*
H40C	0.4971	-0.3382	0.7399	0.103*
C41	0.4530 (4)	0.0235 (4)	0.5799 (3)	0.0356 (12)
H41A	0.4764	0.0657	0.5327	0.043*
H41B	0.4163	-0.0311	0.5741	0.043*
C42	0.3802 (3)	0.0938 (4)	0.6162 (3)	0.0323 (11)
C43	0.3087 (4)	0.0466 (4)	0.6706 (3)	0.0380 (12)
H43	0.3024	-0.0250	0.6812	0.046*
C44	0.2465 (4)	0.1071 (4)	0.7093 (3)	0.0363 (12)
C45	0.2605 (4)	0.2121 (4)	0.6934 (3)	0.0374 (12)
H45	0.2208	0.2516	0.7198	0.045*
C46	0.3309 (3)	0.2633 (4)	0.6397 (2)	0.0333 (11)
C47	0.3896 (3)	0.2003 (4)	0.6010 (2)	0.0319 (11)
C48	0.1656 (4)	0.0584 (5)	0.7686 (3)	0.0496 (15)
C49	0.1701 (7)	-0.0603 (6)	0.7842 (5)	0.092 (3)
H49A	0.1507	-0.0758	0.7432	0.138*
H49B	0.1237	-0.0880	0.8246	0.138*
H49C	0.2385	-0.0922	0.7946	0.138*
C50	0.0580 (5)	0.1101 (7)	0.7444 (4)	0.079 (2)
H50A	0.0553	0.1855	0.7333	0.118*
H50B	0.0073	0.0845	0.7823	0.118*
H50C	0.0446	0.0925	0.7028	0.118*
C51	0.1810 (5)	0.0845 (6)	0.8371 (3)	0.065 (2)
H51A	0.2480	0.0550	0.8529	0.098*
H51B	0.1301	0.0550	0.8737	0.098*

H51C	0.1744	0.1597	0.8277	0.098*	
C52	0.3504 (4)	0.3764 (4)	0.6306 (3)	0.0348 (12)	
H52A	0.2854	0.4196	0.6321	0.042*	
H52B	0.3859	0.4050	0.5843	0.042*	
C53	0.4231 (4)	0.2907 (5)	0.4771 (3)	0.0424 (13)	
H53A	0.3676	0.3462	0.4771	0.051*	
H53B	0.3975	0.2360	0.4630	0.051*	
C54	0.5112 (4)	0.3343 (4)	0.4267 (3)	0.0385 (12)	
H54A	0.4889	0.3668	0.3787	0.046*	
H54B	0.5384	0.3871	0.4420	0.046*	
C55	0.6794 (4)	0.2639 (4)	0.3893 (3)	0.0308 (11)	
C56	0.6960 (4)	0.3571 (4)	0.3360 (3)	0.0470 (15)	
H56	0.6447	0.4135	0.3263	0.056*	
C57	0.7906 (5)	0.3650 (5)	0.2974 (4)	0.0610 (19)	
H57	0.8027	0.4267	0.2611	0.073*	
C58	0.8660 (4)	0.2816 (4)	0.3128 (3)	0.0456 (14)	
C59	0.8509 (4)	0.1895 (4)	0.3659 (3)	0.0406 (13)	
Н59	0.9027	0.1336	0.3756	0.049*	
C60	0.7566 (4)	0.1811 (4)	0.4049 (3)	0.0350 (11)	
H60	0.7456	0.1196	0.4417	0.042*	
C61	1.0401 (8)	0.6203 (8)	0.1833 (6)	0.105 (3)	
H61A	1.0938	0.5636	0.1852	0.157*	
H61B	1.0696	0.6843	0.1777	0.157*	
H61C	0.9998	0.6295	0.1435	0.157*	
C62	0.9756 (6)	0.5953 (7)	0.2489 (6)	0.085 (3)	
C63	0.7033 (10)	0.2776 (8)	1.0540 (4)	0.115 (4)	
H63A	0.7504	0.2147	1.0597	0.173*	
H63B	0.6344	0.2588	1.0645	0.173*	
H63C	0.7179	0.3141	1.0862	0.173*	
C64	0.7141 (6)	0.3450 (6)	0.9816 (4)	0.072 (2)	
C65	0.5154 (4)	0.0915 (5)	0.7801 (3)	0.0500 (15)	
H65A	0.4639	0.1254	0.7469	0.075*	
H65B	0.5748	0.0659	0.7553	0.075*	
H65C	0.5338	0.1413	0.8015	0.075*	
C66	0.4757 (5)	0.0040 (5)	0.8351 (4)	0.0561 (17)	
C67	1.0242 (10)	0.3631 (11)	0.0428 (9)	0.191 (8)	
C68	1.1128 (8)	0.3882 (10)	0.0768 (6)	0.124 (4)	
H68A	1.1753	0.3493	0.0652	0.186*	
H68B	1.0995	0.3686	0.1278	0.186*	
H68C	1.1194	0.4625	0.0587	0.186*	
C1'	0.443 (2)	0.382 (3)	0.9414 (12)	0.080 (11)	0.261 (13)
H1'1	0.4054	0.3841	0.9854	0.120*	0.261 (13)
H1'2	0.4680	0.4496	0.9180	0.120*	0.261 (13)
H1'3	0.5007	0.3285	0.9515	0.120*	0.261 (13)
C2'	0.321 (3)	0.2532 (16)	0.9184 (16)	0.098 (14)	0.261 (13)
H2'1	0.2713	0.2549	0.8857	0.147*	0.261 (13)
H2'2	0.2878	0.2466	0.9651	0.147*	0.261 (13)
H2'3	0.3721	0.1940	0.9208	0.147*	0.261 (13)
C3'	0.2903 (18)	0.4530 (16)	0.8839 (13)	0.067 (9)	0.261 (13)

H3'1	0.3229	0.5170	0.8691	0.101*	0.261 (13)
H3'2	0.2522	0.4472	0.9289	0.101*	0.261 (13)
H3'3	0.2448	0.4545	0.8486	0.101*	0.261 (13)
N1	0.9639 (4)	0.2891 (4)	0.2696 (3)	0.0644 (16)	
N2	0.8937 (4)	0.8818 (4)	0.4872 (3)	0.0566 (14)	
N3	0.9248 (6)	0.5756 (7)	0.3004 (5)	0.114 (3)	
N4	0.4436 (6)	-0.0641 (6)	0.8780 (5)	0.111 (3)	
N5	0.7228 (7)	0.3959 (6)	0.9247 (4)	0.100 (2)	
N6	0.9436 (18)	0.342 (2)	0.031 (2)	0.45 (3)	
01	0.6488 (2)	0.1190 (3)	0.56909 (18)	0.0364 (8)	
H1	0.5930	0.1536	0.5602	0.055*	
O2	0.5649 (3)	0.3939 (3)	0.61067 (17)	0.0381 (9)	
H2	0.6211	0.3589	0.6125	0.057*	
O3	0.4624 (2)	0.2478 (3)	0.54692 (16)	0.0337 (8)	
O4	0.7515 (2)	0.2675 (2)	0.62072 (16)	0.0291 (7)	
O5	0.8632 (3)	0.4473 (3)	0.57552 (17)	0.0365 (8)	
O6	0.5876 (2)	0.2470 (3)	0.42839 (18)	0.0350 (8)	
O7	0.9145 (6)	0.9262 (4)	0.4246 (3)	0.0969 (15)	
O8	0.8850 (6)	0.9283 (4)	0.5324 (3)	0.0969 (15)	
09	1.0274 (4)	0.2139 (5)	0.2778 (4)	0.1034 (16)	
O10	0.9803 (4)	0.3736 (5)	0.2274 (4)	0.1034 (16)	

### Atomic displacement parameters $(\text{\AA}^2)$

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
C1	0.061 (6)	0.108 (9)	0.046 (5)	-0.026 (6)	0.023 (4)	-0.034 (5)
C2	0.102 (10)	0.069 (7)	0.051 (6)	-0.028 (7)	0.031 (6)	-0.031 (6)
C3	0.110 (10)	0.053 (6)	0.042 (5)	-0.007 (5)	0.022 (5)	-0.008 (4)
C4	0.054 (4)	0.062 (4)	0.031 (3)	-0.003 (3)	0.009 (2)	-0.022 (3)
C5	0.044 (3)	0.034 (3)	0.034 (3)	0.004 (2)	0.006 (2)	-0.010 (2)
C6	0.031 (3)	0.037 (3)	0.036 (3)	0.006 (2)	0.003 (2)	-0.007 (2)
C7	0.026 (2)	0.025 (2)	0.037 (3)	0.0058 (19)	-0.002 (2)	-0.008 (2)
C8	0.034 (3)	0.025 (2)	0.027 (2)	0.007 (2)	0.000 (2)	-0.0053 (19)
C9	0.038 (3)	0.022 (2)	0.034 (3)	0.000(2)	-0.001 (2)	-0.006 (2)
C10	0.045 (3)	0.031 (3)	0.033 (3)	0.003 (2)	-0.003 (2)	-0.012 (2)
C11	0.031 (3)	0.033 (3)	0.039 (3)	-0.002 (2)	-0.003 (2)	-0.012 (2)
C12	0.017 (2)	0.031 (2)	0.039 (3)	-0.0042 (19)	-0.0008 (19)	-0.010 (2)
C13	0.018 (2)	0.033 (3)	0.031 (2)	-0.0054 (19)	-0.0015 (18)	-0.008 (2)
C14	0.019 (2)	0.027 (2)	0.038 (3)	-0.0016 (18)	0.0002 (18)	-0.009 (2)
C15	0.022 (2)	0.028 (2)	0.041 (3)	0.0006 (19)	-0.003 (2)	-0.007 (2)
C16	0.025 (2)	0.037 (3)	0.038 (3)	-0.002 (2)	-0.003 (2)	-0.010 (2)
C17	0.024 (2)	0.041 (3)	0.034 (3)	-0.003 (2)	0.0000 (19)	-0.014 (2)
C18	0.036 (3)	0.049 (3)	0.036 (3)	0.004 (2)	-0.005 (2)	-0.008 (2)
C19	0.050 (4)	0.084 (5)	0.038 (3)	-0.007 (3)	0.002 (3)	-0.001 (3)
C20	0.047 (4)	0.079 (5)	0.038 (3)	0.000 (3)	-0.013 (3)	-0.008 (3)
C21	0.096 (6)	0.062 (4)	0.040 (3)	0.016 (4)	-0.016 (3)	0.002 (3)
C22	0.032 (3)	0.035 (3)	0.037 (3)	-0.003 (2)	0.005 (2)	-0.010 (2)
C23	0.033 (3)	0.033 (3)	0.032 (3)	-0.003 (2)	0.000 (2)	-0.012 (2)

C24	0.022 (2)	0.036 (3)	0.038 (3)	-0.006 (2)	0.0006 (19)	-0.014 (2)
C25	0.070 (4)	0.043 (3)	0.033 (3)	-0.014 (3)	0.003 (3)	-0.011 (2)
C26	0.077 (5)	0.054 (4)	0.044 (3)	-0.017 (3)	0.013 (3)	-0.027 (3)
C27	0.035 (3)	0.033 (3)	0.052 (3)	-0.004 (2)	0.001 (2)	-0.018 (2)
C28	0.033 (3)	0.037 (3)	0.038 (3)	-0.005 (2)	-0.008 (2)	-0.006 (2)
C29	0.029 (3)	0.040 (3)	0.035 (3)	-0.006 (2)	-0.004 (2)	-0.014 (2)
C30	0.022 (2)	0.028 (2)	0.038 (3)	0.0025 (19)	-0.0030 (19)	-0.010 (2)
C31	0.026 (2)	0.029 (2)	0.033 (2)	0.0000 (19)	0.0013 (19)	-0.017 (2)
C32	0.027 (2)	0.030 (2)	0.031 (2)	-0.0015 (19)	0.0033 (19)	-0.013 (2)
C33	0.027 (2)	0.037 (3)	0.033 (2)	-0.006 (2)	0.0027 (19)	-0.019 (2)
C34	0.034 (3)	0.038 (3)	0.044 (3)	-0.009 (2)	0.007 (2)	-0.021 (2)
C35	0.041 (3)	0.033 (3)	0.038 (3)	-0.005 (2)	0.007 (2)	-0.016 (2)
C36	0.034 (3)	0.029 (3)	0.033 (3)	0.005 (2)	-0.002 (2)	-0.012 (2)
C37	0.056 (4)	0.029 (3)	0.047 (3)	-0.007 (2)	0.006 (3)	-0.010 (2)
C38	0.080 (5)	0.033 (3)	0.068 (4)	0.000 (3)	0.012 (4)	-0.012 (3)
C39	0.092 (5)	0.046 (4)	0.042 (3)	-0.005 (3)	-0.002 (3)	-0.007 (3)
C40	0.062 (4)	0.057 (4)	0.077 (5)	-0.023 (3)	0.006 (4)	-0.003 (4)
C41	0.029 (3)	0.043 (3)	0.039 (3)	-0.007 (2)	0.000 (2)	-0.019 (2)
C42	0.019 (2)	0.046 (3)	0.033 (3)	-0.004 (2)	-0.0023 (19)	-0.013 (2)
C43	0.029 (3)	0.047 (3)	0.038 (3)	-0.006 (2)	-0.001 (2)	-0.013 (2)
C44	0.021 (2)	0.057 (3)	0.031 (3)	-0.009 (2)	0.0002 (19)	-0.012 (2)
C45	0.019 (2)	0.058 (3)	0.036 (3)	0.002 (2)	0.002 (2)	-0.018 (2)
C46	0.020 (2)	0.048 (3)	0.030 (2)	0.003 (2)	-0.0056 (19)	-0.009 (2)
C47	0.021 (2)	0.048 (3)	0.026 (2)	-0.005 (2)	0.0005 (18)	-0.011 (2)
C48	0.035 (3)	0.078 (4)	0.037 (3)	-0.019 (3)	0.008 (2)	-0.016 (3)
C49	0.105 (7)	0.084 (6)	0.087 (6)	-0.052 (5)	0.052 (5)	-0.026 (5)
C50	0.028 (3)	0.148 (8)	0.057 (4)	-0.021 (4)	0.007 (3)	-0.023 (4)
C51	0.050 (4)	0.108 (6)	0.036 (3)	-0.020 (4)	0.006 (3)	-0.017 (3)
C52	0.020 (2)	0.045 (3)	0.033 (3)	0.012 (2)	-0.0036 (19)	-0.008(2)
C53	0.030 (3)	0.059 (4)	0.030 (3)	0.009 (2)	-0.005 (2)	-0.006 (2)
C54	0.029 (3)	0.044 (3)	0.035 (3)	0.011 (2)	-0.004 (2)	-0.007 (2)
C55	0.025 (2)	0.032 (3)	0.036 (3)	0.002 (2)	-0.0001 (19)	-0.013 (2)
C56	0.042 (3)	0.034 (3)	0.054 (3)	0.011 (2)	0.009 (3)	-0.009 (3)
C57	0.056 (4)	0.043 (3)	0.067 (4)	0.001 (3)	0.022 (3)	-0.002 (3)
C58	0.031 (3)	0.043 (3)	0.061 (4)	-0.002 (2)	0.013 (2)	-0.020 (3)
C59	0.034 (3)	0.037 (3)	0.054 (3)	0.001 (2)	-0.003 (2)	-0.020 (3)
C60	0.028 (3)	0.036 (3)	0.039 (3)	0.001 (2)	-0.004 (2)	-0.009 (2)
C61	0.084 (6)	0.100 (7)	0.130 (8)	-0.020 (5)	0.011 (6)	-0.034 (6)
C62	0.047 (4)	0.087 (6)	0.120 (8)	0.010 (4)	-0.001 (5)	-0.037 (5)
C63	0.185 (12)	0.094 (7)	0.062 (5)	-0.035 (7)	0.026 (6)	-0.020 (5)
C64	0.081 (5)	0.066 (5)	0.069 (5)	-0.011 (4)	0.007 (4)	-0.025 (4)
C65	0.035 (3)	0.059 (4)	0.054 (3)	0.001 (3)	-0.002 (3)	-0.017 (3)
C66	0.036 (3)	0.056 (4)	0.073 (4)	0.006 (3)	0.008 (3)	-0.023 (3)
C67	0.129 (12)	0.125 (11)	0.26 (2)	0.031 (10)	-0.045 (13)	0.016 (12)
C68	0.087 (7)	0.176 (11)	0.104 (8)	0.013 (7)	-0.002 (6)	-0.048 (8)
C1'	0.13 (3)	0.09 (3)	0.021 (13)	-0.01 (2)	0.019 (15)	-0.026 (16)
C2'	0.08 (2)	0.12 (3)	0.060 (18)	0.01 (2)	0.017 (17)	0.016 (18)
C3'	0.058 (17)	0.10 (2)	0.053 (15)	0.006 (15)	0.007 (12)	-0.042 (15)
N1	0.055 (3)	0.051 (3)	0.078 (4)	-0.009 (3)	0.030 (3)	-0.015 (3)

N2	0.060 (3)	0.047 (3)	0.069 (4)	-0.007 (2)	0.010 (3)	-0.031 (3)
N3	0.069 (5)	0.131 (7)	0.135 (7)	0.018 (5)	0.009 (5)	-0.045 (6)
N4	0.083 (5)	0.080 (5)	0.131 (7)	0.005 (4)	0.042 (5)	0.002 (5)
N5	0.127 (7)	0.095 (6)	0.071 (5)	-0.016 (5)	-0.011 (4)	-0.012 (4)
N6	0.32 (3)	0.28 (2)	0.72 (6)	-0.06 (2)	-0.35 (4)	0.03 (3)
01	0.0257 (18)	0.0349 (19)	0.044 (2)	-0.0021 (14)	-0.0061 (15)	-0.0038 (16)
02	0.0300 (19)	0.047 (2)	0.0323 (18)	0.0081 (16)	0.0003 (14)	-0.0103 (16)
O3	0.0242 (17)	0.045 (2)	0.0286 (17)	-0.0003 (15)	0.0019 (13)	-0.0086 (15)
O4	0.0203 (16)	0.0319 (17)	0.0329 (17)	0.0002 (13)	0.0004 (13)	-0.0083 (14)
05	0.043 (2)	0.0330 (19)	0.0336 (18)	-0.0107 (16)	-0.0014 (15)	-0.0078 (15)
O6	0.0249 (17)	0.0347 (19)	0.0413 (19)	0.0031 (14)	0.0014 (14)	-0.0092 (15)
O7	0.169 (5)	0.051 (2)	0.073 (2)	-0.028 (2)	0.019 (3)	-0.0238 (19)
08	0.169 (5)	0.051 (2)	0.073 (2)	-0.028 (2)	0.019 (3)	-0.0238 (19)
09	0.063 (2)	0.080 (3)	0.137 (4)	0.007 (2)	0.051 (2)	-0.014 (3)
O10	0.063 (2)	0.080 (3)	0.137 (4)	0.007 (2)	0.051 (2)	-0.014 (3)

Geometric parameters (Å, °)

C1—C4	1.580 (11)	C38—H38B	0.9600
C1—H1A	0.9600	C38—H38C	0.9600
C1—H1B	0.9600	C39—H39A	0.9600
C1—H1C	0.9600	С39—Н39В	0.9600
C2—C4	1.475 (12)	С39—Н39С	0.9600
C2—H2A	0.9600	C40—H40A	0.9600
C2—H2B	0.9600	C40—H40B	0.9600
C2—H2C	0.9600	C40—H40C	0.9600
C3—C4	1.550 (11)	C41—C42	1.532 (7)
С3—НЗА	0.9600	C41—H41A	0.9700
С3—Н3В	0.9600	C41—H41B	0.9700
С3—НЗС	0.9600	C42—C47	1.384 (7)
C4—C5	1.548 (7)	C42—C43	1.404 (7)
C4—C2'	1.5498 (11)	C43—C44	1.409 (8)
C4—C1'	1.5501 (11)	C43—H43	0.9300
C4—C3'	1.5501 (11)	C44—C45	1.376 (8)
C5—C6	1.390 (8)	C44—C48	1.543 (7)
C5—C10	1.415 (8)	C45—C46	1.403 (7)
C6—C7	1.400 (7)	C45—H45	0.9300
С6—Н6	0.9300	C46—C47	1.411 (7)
С7—С8	1.408 (7)	C46—C52	1.523 (7)
C7—C52	1.531 (7)	C47—O3	1.412 (5)
C8—O2	1.377 (5)	C48—C49	1.517 (10)
C8—C9	1.400 (7)	C48—C51	1.534 (9)
C9—C10	1.396 (7)	C48—C50	1.547 (9)
C9—C11	1.532 (7)	C49—H49A	0.9600
C10—H10	0.9300	C49—H49B	0.9600
C11—C12	1.524 (7)	C49—H49C	0.9600
C11—H11A	0.9700	C50—H50A	0.9600
C11—H11B	0.9700	C50—H50B	0.9600
C12—C17	1.367 (7)	С50—Н50С	0.9600

C12—C13	1.422 (7)	C51—H51A	0.9600
C13—O4	1.386 (5)	C51—H51B	0.9600
C13—C14	1.407 (7)	C51—H51C	0.9600
C14—C15	1.388 (7)	C52—H52A	0.9700
C14—C30	1.532 (7)	C52—H52B	0.9700
C15—C16	1.400 (7)	C53—O3	1.438 (6)
C15—H15	0.9300	C53—C54	1.504 (7)
C16—C17	1.413 (7)	С53—Н53А	0.9700
C16—C18	1.529 (7)	С53—Н53В	0.9700
С17—Н17	0.9300	C54—O6	1.431 (6)
C18—C21	1.524 (9)	C54—H54A	0.9700
C18—C19	1.544 (8)	С54—Н54В	0.9700
C18—C20	1.567 (9)	C55—O6	1.372 (5)
C19—H19A	0.9600	C55—C60	1.382 (7)
C19—H19B	0.9600	C55—C56	1.387 (7)
С19—Н19С	0.9600	C56—C57	1.392 (8)
C20—H20A	0.9600	С56—Н56	0.9300
С20—Н20В	0.9600	C57—C58	1.371 (8)
С20—Н20С	0.9600	С57—Н57	0.9300
C21—H21A	0.9600	C58—C59	1.371 (8)
C21—H21B	0.9600	C58—N1	1.467 (7)
C21—H21C	0.9600	C59—C60	1.391 (7)
C22—O4	1.443 (5)	С59—Н59	0.9300
C22—C23	1.505 (7)	С60—Н60	0.9300
C22—H22A	0.9700	C61—C62	1.444 (13)
C22—H22B	0.9700	C61—H61A	0.9600
C23—O5	1.434 (6)	С61—Н61В	0.9600
C23—H23A	0.9700	C61—H61C	0.9600
С23—Н23В	0.9700	C62—N3	1.133 (11)
C24—O5	1.360 (6)	C63—C64	1.436 (11)
C24—C29	1.385 (7)	С63—Н63А	0.9600
C24—C25	1.405 (8)	С63—Н63В	0.9600
C25—C26	1.378 (8)	С63—Н63С	0.9600
C25—H25	0.9300	C64—N5	1.121 (9)
C26—C27	1.383 (8)	C65—C66	1.443 (9)
С26—Н26	0.9300	С65—Н65А	0.9600
C27—C28	1.383 (8)	С65—Н65В	0.9600
C27—N2	1.463 (7)	С65—Н65С	0.9600
C28—C29	1.378 (7)	C66—N4	1.131 (9)
C28—H28	0.9300	C67—N6	1.20 (3)
С29—Н29	0.9300	C67—C68	1.529 (18)
C30—C31	1.542 (7)	C68—H68A	0.9600
C30—H30A	0.9700	С68—Н68В	0.9600
C30—H30B	0.9700	C68—H68C	0.9600
C31—C36	1.377 (7)	С1'—Н1'1	0.9600
C31—C32	1.417 (7)	C1'—H1'2	0.9600
C32—O1	1.369 (5)	С1'—Н1'3	0.9600
C32—C33	1.396 (7)	C2'—H2'1	0.9600
C33—C34	1.390 (7)	C2'—H2'2	0.9600

C33—C41	1.545 (7)	C2'—H2'3	0.9600
C34—C35	1.390 (8)	С3'—Н3'1	0.9600
С34—Н34	0.9300	C3'—H3'2	0.9600
C35—C36	1.417 (7)	C3'—H3'3	0.9600
C35—C37	1.528 (7)	N1—O9	1.202 (7)
С36—Н36	0.9300	N1—O10	1.210 (7)
C37—C38	1.534 (8)	N2—O7	1.212 (7)
C37—C39	1.540 (9)	N2—O8	1.222 (7)
C37—C40	1.546 (9)	O1—H1	0.8200
C38—H38A	0.9600	O2—H2	0.8200
C4—C1—H1A	109.5	H38A—C38—H38C	109.5
C4—C1—H1B	109.5	H38B—C38—H38C	109.5
H1A—C1—H1B	109.5	С37—С39—Н39А	109.5
C4—C1—H1C	109.5	С37—С39—Н39В	109.5
H1A—C1—H1C	109.5	H39A—C39—H39B	109.5
H1B—C1—H1C	109.5	С37—С39—Н39С	109.5
C4—C2—H2A	109.5	Н39А—С39—Н39С	109.5
C4—C2—H2B	109.5	H39B—C39—H39C	109.5
H2A—C2—H2B	109.5	С37—С40—Н40А	109.5
C4—C2—H2C	109.5	С37—С40—Н40В	109.5
H2A—C2—H2C	109.5	H40A—C40—H40B	109.5
H2B—C2—H2C	109.5	С37—С40—Н40С	109.5
С4—С3—Н3А	109.5	H40A—C40—H40C	109.5
С4—С3—Н3В	109.5	H40B-C40-H40C	109.5
НЗА—СЗ—НЗВ	109.5	C42—C41—C33	111.6 (4)
С4—С3—Н3С	109.5	C42—C41—H41A	109.3
НЗА—СЗ—НЗС	109.5	C33—C41—H41A	109.3
НЗВ—СЗ—НЗС	109.5	C42—C41—H41B	109.3
C2—C4—C5	112.3 (6)	С33—С41—Н41В	109.3
C2—C4—C3	111.4 (8)	H41A—C41—H41B	108.0
C5—C4—C3	107.2 (5)	C47—C42—C43	119.5 (5)
C2'—C4—C1'	119.7 (19)	C47—C42—C41	121.5 (4)
C2'—C4—C3'	109.6 (18)	C43—C42—C41	118.7 (5)
C1'—C4—C3'	99 (2)	C42—C43—C44	120.4 (5)
C2—C4—C1	112.0 (8)	C42—C43—H43	119.8
C5—C4—C1	111.0 (5)	C44—C43—H43	119.8
C3—C4—C1	102.3 (7)	C45—C44—C43	117.8 (4)
C6—C5—C10	117.0 (5)	C45—C44—C48	120.3 (5)
C6—C5—C4	122.9 (5)	C43—C44—C48	121.8 (5)
C10—C5—C4	120.1 (5)	C44—C45—C46	124.2 (5)
C5—C6—C7	122.9 (5)	C44—C45—H45	117.9
С5—С6—Н6	118.6	C46—C45—H45	117.9
С7—С6—Н6	118.6	C45—C46—C47	116.0 (5)
C6—C7—C8	118.5 (5)	C45—C46—C52	121.1 (5)
C6—C7—C52	119.7 (4)	C47—C46—C52	122.6 (4)
C8—C7—C52	121.6 (4)	C42—C47—C46	122.1 (4)
O2—C8—C9	122.2 (4)	C42—C47—O3	119.6 (4)
O2—C8—C7	117.2 (4)	C46—C47—O3	118.3 (4)
C9—C8—C7	120.6 (4)	C49—C48—C51	109.7 (6)

C10—C9—C8	119.0 (5)	C49—C48—C44	112.5 (5)
C10-C9-C11	118.9 (5)	C51—C48—C44	109.5 (5)
C8—C9—C11	121.9 (4)	C49—C48—C50	109.1 (6)
C9—C10—C5	122.1 (5)	C51—C48—C50	107.3 (6)
С9—С10—Н10	118.9	C44—C48—C50	108.6 (5)
С5—С10—Н10	118.9	C48—C49—H49A	109.5
C12—C11—C9	110.4 (4)	C48—C49—H49B	109.5
C12—C11—H11A	109.6	H49A—C49—H49B	109.5
С9—С11—Н11А	109.6	C48—C49—H49C	109.5
C12—C11—H11B	109.6	H49A—C49—H49C	109.5
С9—С11—Н11В	109.6	H49B—C49—H49C	109.5
H11A—C11—H11B	108.1	C48—C50—H50A	109.5
C17—C12—C13	119.6 (4)	C48—C50—H50B	109.5
C17—C12—C11	118.9 (5)	H50A—C50—H50B	109.5
C13—C12—C11	121.2 (4)	C48—C50—H50C	109.5
O4—C13—C14	122.8 (4)	H50A—C50—H50C	109.5
O4—C13—C12	116.9 (4)	H50B—C50—H50C	109.5
C14—C13—C12	120.3 (4)	C48—C51—H51A	109.5
C15—C14—C13	117.5 (4)	C48—C51—H51B	109.5
C15—C14—C30	118.2 (4)	H51A—C51—H51B	109.5
C13—C14—C30	124.0 (4)	C48—C51—H51C	109.5
C14—C15—C16	123.9 (5)	H51A—C51—H51C	109.5
C14—C15—H15	118.0	H51B—C51—H51C	109.5
С16—С15—Н15	118.0	C46—C52—C7	111.0 (4)
C15—C16—C17	116.4 (4)	C46—C52—H52A	109.4
C15—C16—C18	124.0 (5)	C7—C52—H52A	109.4
C17—C16—C18	119.5 (5)	C46—C52—H52B	109.4
C12—C17—C16	122.1 (5)	С7—С52—Н52В	109.4
C12—C17—H17	118.9	H52A—C52—H52B	108.0
С16—С17—Н17	118.9	O3—C53—C54	106.3 (4)
C21—C18—C16	112.0 (5)	O3—C53—H53A	110.5
C21—C18—C19	109.1 (5)	С54—С53—Н53А	110.5
C16—C18—C19	109.3 (4)	O3—C53—H53B	110.5
C21—C18—C20	108.3 (5)	С54—С53—Н53В	110.5
C16—C18—C20	109.2 (4)	Н53А—С53—Н53В	108.7
C19—C18—C20	109.0 (5)	O6—C54—C53	106.7 (4)
С18—С19—Н19А	109.5	O6—C54—H54A	110.4
C18—C19—H19B	109.5	C53—C54—H54A	110.4
H19A—C19—H19B	109.5	O6—C54—H54B	110.4
C18—C19—H19C	109.5	C53—C54—H54B	110.4
H19A—C19—H19C	109.5	H54A—C54—H54B	108.6
H19B—C19—H19C	109.5	O6—C55—C60	116.3 (4)
C18—C20—H20A	109.5	O6—C55—C56	123.3 (4)
C18—C20—H20B	109.5	C60—C55—C56	120.4 (4)
H20A—C20—H20B	109.5	C55—C56—C57	119.0 (5)
C18—C20—H20C	109.5	С55—С56—Н56	120.5
H20A—C20—H20C	109.5	С57—С56—Н56	120.5
H20B—C20—H20C	109.5	C58—C57—C56	120.0 (5)
C18—C21—H21A	109.5	С58—С57—Н57	120.0

C18—C21—H21B	109.5	С56—С57—Н57	120.0
H21A—C21—H21B	109.5	C57—C58—C59	121.5 (5)
C18—C21—H21C	109.5	C57—C58—N1	119.5 (5)
H21A—C21—H21C	109.5	C59—C58—N1	118.9 (5)
H21B-C21-H21C	109.5	C58—C59—C60	118.9 (5)
O4—C22—C23	109.3 (4)	С58—С59—Н59	120.6
O4—C22—H22A	109.8	С60—С59—Н59	120.6
C23—C22—H22A	109.8	C55—C60—C59	120.2 (5)
O4—C22—H22B	109.8	С55—С60—Н60	119.9
С23—С22—Н22В	109.8	С59—С60—Н60	119.9
H22A—C22—H22B	108.3	С62—С61—Н61А	109.5
O5—C23—C22	108.5 (4)	С62—С61—Н61В	109.5
O5—C23—H23A	110.0	H61A—C61—H61B	109.5
С22—С23—Н23А	110.0	С62—С61—Н61С	109.5
O5—C23—H23B	110.0	H61A—C61—H61C	109.5
С22—С23—Н23В	110.0	H61B—C61—H61C	109.5
H23A—C23—H23B	108.4	N3—C62—C61	179.9 (13)
O5—C24—C29	124.8 (5)	С64—С63—Н63А	109.5
O5—C24—C25	116.2 (4)	С64—С63—Н63В	109.5
C29—C24—C25	119.0 (5)	Н63А—С63—Н63В	109.5
C26—C25—C24	121.3 (5)	С64—С63—Н63С	109.5
С26—С25—Н25	119.3	Н63А—С63—Н63С	109.5
С24—С25—Н25	119.3	H63B—C63—H63C	109.5
C25—C26—C27	118.7 (6)	N5—C64—C63	178.7 (9)
С25—С26—Н26	120.7	С66—С65—Н65А	109.5
С27—С26—Н26	120.7	С66—С65—Н65В	109.5
C28—C27—C26	120.5 (5)	Н65А—С65—Н65В	109.5
C28—C27—N2	120.2 (5)	С66—С65—Н65С	109.5
C26—C27—N2	119.4 (5)	Н65А—С65—Н65С	109.5
C29—C28—C27	121.0 (5)	H65B—C65—H65C	109.5
C29—C28—H28	119.5	N4—C66—C65	179.3 (8)
С27—С28—Н28	119.5	N6—C67—C68	165 (3)
C28—C29—C24	119.5 (5)	С67—С68—Н68А	109.5
С28—С29—Н29	120.2	С67—С68—Н68В	109.5
С24—С29—Н29	120.2	H68A—C68—H68B	109.5
C14—C30—C31	111.1 (4)	С67—С68—Н68С	109.5
С14—С30—Н30А	109.4	H68A—C68—H68C	109.5
C31—C30—H30A	109.4	H68B—C68—H68C	109.5
С14—С30—Н30В	109.4	C4—C1'—H1'1	109.5
С31—С30—Н30В	109.4	C4—C1'—H1'2	109.5
H30A—C30—H30B	108.0	H1'1—C1'—H1'2	109.5
C36—C31—C32	118.1 (4)	C4—C1'—H1'3	109.5
C36—C31—C30	121.3 (4)	H1'1—C1'—H1'3	109.5
C32—C31—C30	120.4 (4)	H1'2—C1'—H1'3	109.5
O1—C32—C33	124.6 (4)	C4—C2'—H2'1	109.5
O1—C32—C31	115.0 (4)	C4—C2'—H2'2	109.5
C33—C32—C31	120.4 (4)	H2'1—C2'—H2'2	109.5
C34—C33—C32	119.5 (5)	C4—C2'—H2'3	109.5
C34—C33—C41	118.5 (4)	H2'1—C2'—H2'3	109.5

C32—C33—C41	121.9 (4)	H2'2—C2'—H2'3	109.5
C35—C34—C33	121.9 (5)	C4—C3'—H3'1	109.5
С35—С34—Н34	119.1	C4—C3'—H3'2	109.5
С33—С34—Н34	119.1	H3'1—C3'—H3'2	109.5
C34—C35—C36	117.2 (5)	C4—C3'—H3'3	109.5
C34—C35—C37	123.2 (5)	H3'1—C3'—H3'3	109.5
C36—C35—C37	119.6 (5)	H3'2—C3'—H3'3	109.5
C31—C36—C35	122.8 (5)	O9—N1—O10	120.8 (5)
С31—С36—Н36	118.6	O9—N1—C58	120.6 (5)
С35—С36—Н36	118.6	O10-N1-C58	118.5 (5)
C35—C37—C38	109.5 (4)	O7—N2—O8	121.7 (6)
C35—C37—C39	109.6 (5)	O7—N2—C27	118.6 (5)
C38—C37—C39	108.9 (5)	O8—N2—C27	119.7 (5)
C35—C37—C40	112.1 (5)	С32—О1—Н1	109.5
C38—C37—C40	108.4 (5)	C8—O2—H2	109.5
C39—C37—C40	108.2 (5)	C47—O3—C53	113.8 (4)
С37—С38—Н38А	109.5	C13—O4—C22	117.7 (3)
С37—С38—Н38В	109.5	C24—O5—C23	119.7 (4)
H38A—C38—H38B	109.5	C55—O6—C54	119.8 (4)
С37—С38—Н38С	109.5		
C2—C4—C5—C6	-123.5 (9)	C31—C32—C33—C41	178.4 (4)
C3—C4—C5—C6	113.8 (7)	C32—C33—C34—C35	-1.5 (7)
C2'—C4—C5—C6	59.9 (16)	C41—C33—C34—C35	-178.0 (5)
C1'—C4—C5—C6	-164.7 (18)	C33—C34—C35—C36	-0.4 (7)
C3'—C4—C5—C6	-57.7 (14)	C33—C34—C35—C37	179.1 (5)
C1—C4—C5—C6	2.8 (8)	C32—C31—C36—C35	-1.3 (7)
C2—C4—C5—C10	58.5 (9)	C30-C31-C36-C35	173.5 (4)
C3—C4—C5—C10	-64.2 (8)	C34—C35—C36—C31	1.9 (7)
C2'—C4—C5—C10	-118.1 (15)	C37—C35—C36—C31	-177.7 (4)
C1'C4C5C10	17.3 (19)	C34—C35—C37—C38	116.4 (6)
C3'—C4—C5—C10	124.2 (13)	C36—C35—C37—C38	-64.1 (7)
C1—C4—C5—C10	-175.2 (6)	C34—C35—C37—C39	-124.1 (6)
C10—C5—C6—C7	0.8 (8)	C36—C35—C37—C39	55.4 (7)
C4—C5—C6—C7	-177.3 (5)	C34—C35—C37—C40	-3.9 (8)
C5—C6—C7—C8	-0.7 (7)	C36—C35—C37—C40	175.6 (5)
C5—C6—C7—C52	174.2 (5)	C34—C33—C41—C42	90.1 (5)
C6—C7—C8—O2	-177.2 (4)	C32—C33—C41—C42	-86.3 (5)
C52—C7—C8—O2	8.0 (7)	C33—C41—C42—C47	91.5 (5)
C6—C7—C8—C9	0.0 (7)	C33—C41—C42—C43	-82.6 (5)
C52—C7—C8—C9	-174.8 (4)	C47—C42—C43—C44	0.3 (7)
O2—C8—C9—C10	177.6 (4)	C41—C42—C43—C44	174.6 (4)
C7—C8—C9—C10	0.5 (7)	C42—C43—C44—C45	-1.9 (7)
O2—C8—C9—C11	-7.8 (7)	C42—C43—C44—C48	178.8 (5)
C7—C8—C9—C11	175.1 (4)	C43—C44—C45—C46	1.8 (8)
C8—C9—C10—C5	-0.3 (7)	C48—C44—C45—C46	-178.9 (5)
C11—C9—C10—C5	-175.1 (4)	C44—C45—C46—C47	-0.1 (7)
C6—C5—C10—C9	-0.3 (7)	C44—C45—C46—C52	-173.2 (4)
C4—C5—C10—C9	177.9 (5)	C43—C42—C47—C46	1.4 (7)
C10—C9—C11—C12	86.4 (5)	C41—C42—C47—C46	-172.7 (4)

C8—C9—C11—C12	-88.2 (5)	C43—C42—C47—O3	179.2 (4)
C9—C11—C12—C17	-82.2 (5)	C41—C42—C47—O3	5.1 (7)
C9—C11—C12—C13	90.9 (5)	C45—C46—C47—C42	-1.5 (7)
C17—C12—C13—O4	178.8 (4)	C52—C46—C47—C42	171.5 (4)
C11—C12—C13—O4	5.7 (6)	C45—C46—C47—O3	-179.3 (4)
C17—C12—C13—C14	0.0 (7)	C52—C46—C47—O3	-6.3 (7)
C11—C12—C13—C14	-173.0 (4)	C45—C44—C48—C49	-173.3 (6)
O4—C13—C14—C15	-179.0 (4)	C43—C44—C48—C49	6.0 (8)
C12—C13—C14—C15	-0.4 (6)	C45—C44—C48—C51	-51.1 (7)
O4—C13—C14—C30	-5.2 (7)	C43—C44—C48—C51	128.2 (6)
C12—C13—C14—C30	173.5 (4)	C45—C44—C48—C50	65.8 (7)
C13—C14—C15—C16	-10(7)	C43—C44—C48—C50	-1149(6)
$C_{30}$ $-C_{14}$ $-C_{15}$ $-C_{16}$	-1752(4)	C45-C46-C52-C7	74 4 (5)
$C_{14}$ $C_{15}$ $C_{16}$ $C_{17}$	2,5(7)	C47 - C46 - C52 - C7	-983(5)
$C_{14}$ $C_{15}$ $C_{16}$ $C_{18}$	-179.6(4)	C6-C7-C52-C46	-78.2(5)
$C_{13}$ $C_{12}$ $C_{17}$ $C_{16}$	16(7)	C8 - C7 - C52 - C46	96.5 (5)
$C_{11} - C_{12} - C_{17} - C_{16}$	174 9 (4)	03 - 053 - 054 - 06	62 0 (6)
$C_{15}$ $C_{16}$ $C_{17}$ $C_{12}$ $C_{12}$	-28(7)	06 - C55 - C56 - C57	-1773(6)
$C_{18}$ $C_{16}$ $C_{17}$ $C_{12}$ $C_{12}$	2.0(7)	C60 - C55 - C56 - C57	16(9)
$C_{15}$ $C_{16}$ $C_{18}$ $C_{21}$	5.0.(8)	$C_{55} = C_{56} = C_{57} = C_{58}$	-0.7(10)
$C_{17}$ $C_{16}$ $C_{18}$ $C_{21}$	-177.2(5)	$C_{55} = C_{57} = C_{58} = C_{59}$	0.7(10)
$C_{15}$ $C_{16}$ $C_{18}$ $C_{19}$	-1160(6)	C56-C57-C58-N1	1774(6)
$C_{13} = C_{10} = C_{13} = C_{13}$	61.9(7)	$C_{50} - C_{50} - C_{50} - C_{60}$	-0.2(0)
$C_{17} = C_{10} = C_{13} = C_{17}$	124.9(5)	N1 = C58 = C59 = C60	-177.6(6)
$C_{13} = C_{10} = C_{13} = C_{20}$	-57.3(6)	06 - C55 - C60 - C59	177.0(0)
04 - 022 - 023 - 05	57.5 (0) 65.7 (5)	$C_{5} = C_{5} = C_{6} = C_{5}$	-1.8(8)
04 - 022 - 025 - 03	-170.5(5)	$C_{50} = C_{50} = C_{60} = C_{55}$	1.0(0)
$C_{24} = C_{25} = C_{20}$	-0.8(0)	$C_{58} = C_{59} = C_{60} = C_{55}$	-173 1 (7)
$C_{2}^{2} = C_{2}^{2} = C_{2$	-0.2(10)	$C_{57} = C_{58} = N_1 = O_7$	173.1(7)
$C_{24} = C_{23} = C_{20} = C_{27}$	-0.1(0)	$C_{59} = C_{58} = N_1 = 0_9$	4.3(10)
$C_{23} = C_{20} = C_{27} = C_{28}$	-0.1(9)	$C_{57} = C_{58} = N_1 = O_{10}$	9.2(10)
$C_{23} - C_{20} - C_{27} - N_{2}$	-1/8.8(0)	$C_{39} = C_{38} = N_1 = O_{10}$	-1/5.4(7)
$C_{20} - C_{27} - C_{28} - C_{29}$	-170.0(5)	$C_{26} = C_{27} = N_2 = O_7$	-9.7(9)
$N_2 = C_2 7 = C_2 8 = C_2 9$	-1/9.9(3) -2.5(7)	$C_{20} = C_{27} = N_{2} = O_{7}$	109.0(7) 172.7(6)
$C_2/-C_{28}-C_{29}-C_{24}$	-2.3(7)	$C_{26} = C_{27} = N_{2} = O_{8}$	1/5.7(0)
03 - 024 - 029 - 028	-1/9.5(3)	$C_{20} = C_{27} = N_{2} = 0.08$	-7.0(9)
$C_{25} - C_{24} - C_{29} - C_{28}$	2.1(7)	C42 - C47 - O3 - C53	90.0 (5)
$C_{13} = C_{14} = C_{30} = C_{31}$	/4.8 (5)	$C_{40} - C_{47} - O_{3} - C_{33}$	-91.0 (5)
$C_{13} - C_{14} - C_{30} - C_{31}$	-99.0 (5)	$C_{34} = C_{33} = 0_{3} = C_{47}$	-1/8.3(4)
C14 - C30 - C31 - C30	-85.5(5)	C14 - C13 - O4 - C22	-62.0(6)
C14 - C30 - C31 - C32	89.2 (5)	C12 - C13 - O4 - C22	119.3 (4)
$C_{30} = C_{31} = C_{32} = C_{11}$	-1/9./(4)	$C_{23} - C_{22} - C_{4} - C_{13}$	-133.4(4)
$C_{30} = C_{31} = C_{32} = C_{32}$	<b>5.5 (6)</b>	$C_{29} = C_{24} = C_{23}$	-9.1 (/)
$C_{30} = C_{31} = C_{32} = C_{33}$	-0./(/)	$C_{23} = C_{24} = C_{23} = C_{23} = C_{24} = C_{23} = C_{24} = C$	109.5 (5)
$C_{30} - C_{31} - C_{32} - C_{33}$	-1/5.6(4)	$C_{22} - C_{23} - C_{5} - C_{24}$	180.0 (4)
U1 - U32 - U33 - U34	-1/9.1(4)	$C_{00}$ $-C_{55}$ $-O_{6}$ $-C_{54}$	166.9 (4)
C31—C32—C33—C34	2.1 (7)	C56—C55—O6—C54	-14.2 (7)
O1—C32—C33—C41	-2.7 (7)	C53—C54—O6—C55	-177.4 (4)

## Hydrogen-bond geometry (Å, °)

D—H···A	<i>D</i> —Н	H···A	$D \cdots A$	D—H··· $A$
01—H1···O3	0.82	1.99	2.802 (6)	171
O2—H2…O4	0.82	1.97	2.781 (4)	176
C50—H50C····O7 <sup>i</sup>	0.96	2.59	3.487 (10)	156
C68—H68B…O10	0.96	2.39	3.247 (13)	148
C65—H65A···Cg1	0.96	2.66	3.590 (6)	163
Symmetry codes: (i) $-x+1, -y+1, -z+1$ .				

Fig. 1

